



## Review

# Some Photosystem II properties depending on the D1 protein variants in *Thermosynechococcus elongatus*<sup>☆</sup>


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## ABSTRACT

Cyanobacteria have multiple *psbA* genes encoding PsbA, the D1 reaction center protein of the Photosystem II complex which bears together with PsbD, the D2 protein, most of the cofactors involved in electron transfer reactions. The thermophilic cyanobacterium *Thermosynechococcus elongatus* has three *psbA* genes differently expressed depending on the environmental conditions. Among the 344 residues constituting each of the 3 possible PsbA variants there are 21 substitutions between PsbA1 and PsbA3, 31 between PsbA1 and PsbA2 and 27 between PsbA2 and PsbA3. In this review, we summarize the changes already identified in the properties of the redox cofactors depending on the D1 variant constituting Photosystem II in *T. elongatus*. This article is part of a Special Issue entitled: Photosynthesis Research for Sustainability: Keys to Produce Clean Energy.

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## 1. Introduction

The light-driven oxidation of water in Photosystem II (PSII) is the first step in the photosynthetic production of most of biomass, fossil fuels and O<sub>2</sub> on Earth. PSII in cyanobacteria is made up of 17 membrane protein subunits and 3 extrinsic proteins (PsbY was not detected in [1] but seen in [2]). Altogether these bear 35 chlorophylls (Chl), 2 pheophytins (Phe), 2 hemes, 1 non-heme iron, 2 plastoquinones (Q<sub>A</sub> and Q<sub>B</sub>), a Mn<sub>4</sub>CaO<sub>5</sub> cluster, at least 2 Cl<sup>−</sup>, 12 carotenoids and 25 lipids [1]. The excitation resulting from the absorption of a photon is transferred to the photochemical trap that undergoes

charge separation. The positive charge is then stabilized on P<sub>680</sub> which is composed of four Chl<sub>a</sub> molecules, P<sub>D1</sub>/P<sub>D2</sub> and Chl<sub>D1</sub>/Chl<sub>D2</sub>, and two pheophytin *a* molecules, Phe<sub>D1</sub>/Phe<sub>D2</sub>. Then, P<sub>680</sub><sup>++</sup> oxidizes Tyr<sub>Z</sub>, the Tyr161 of the D1 polypeptide, which in turn oxidizes the Mn<sub>4</sub>CaO<sub>5</sub> cluster. On the electron acceptor side the electron is transferred to the primary quinone electron acceptor, Q<sub>A</sub>, and then to Q<sub>B</sub>, a two-electron and two-proton acceptor, e.g. [3–5]. The Mn<sub>4</sub>CaO<sub>5</sub> cluster both accumulates oxidizing equivalents and acts as the catalytic site for water oxidation. The enzyme cycles sequentially through five redox states denoted S<sub>*n*</sub> where *n* stands for the number of stored oxidizing equivalents. Upon formation of the S<sub>4</sub> state two molecules of water are rapidly oxidized; the S<sub>0</sub> state is regenerated and O<sub>2</sub> is released [4–8].

Cyanobacterial species have multiple *psbA* variants coding for the D1 protein, e.g. [9–17]. These different genes are known to be differentially expressed depending on the environmental conditions, e.g. [9–15]. In particular, specific up/down-regulation of one of these genes under high light conditions is indicative of a photo-protection mechanism. For example [12], the mesophilic cyanobacterium, *Synechocystis* PCC 6803, has three *psbA* genes. Two of these (*psbAII* and *psbAIII*) produce an identical D1. Nevertheless, while *psbAII* is expressed under the “normal” cultivation conditions, transcription of *psbAIII* is induced by high light or UV light [12]. The expression of *psbAI* seems triggered by micro-aerobic conditions [15]. Other cryptic *psbA* genes were found to be induced in micro-aerobic conditions [18,19]. A class of “rogue” D1 protein has also been recently described [20]. Recent reviews on the

**Abbreviations:** PSII, Photosystem II; Q<sub>A</sub>, primary quinone acceptor; Q<sub>B</sub>, secondary quinone acceptor; Chl, chlorophyll; P<sub>680</sub>, chlorophyll dimer acting as the second electron donor; P<sub>D1</sub> and P<sub>D2</sub>, Chl monomer of P<sub>680</sub> on the D1 or D2 side, respectively; Chl<sub>D1</sub> and Chl<sub>D2</sub>, accessory Chl on the D1 or D2 side, respectively; Phe<sub>D1</sub> and Phe<sub>D2</sub>, pheophytin on the D1 or D2 side, respectively; Tyr<sub>Z</sub>, the Tyr161 of the D1 polypeptide; EPR, Electron Paramagnetic Resonance; SQDG, sulfoquinovosyldiacylglycerol; DCMU, 3-(3,4-dichlorophenyl)-1,1-dimethylurea; PPBQ, phenyl-*p*-benzoquinone; SDS, sodium dodecyl sulfate; MALDI-TOF, matrix-assisted laser desorption/ionization-time of flight; WT\*1, WT\*2, WT\*3, cells containing only the *psbA*<sub>1</sub>, *psbA*<sub>2</sub>, and *psbA*<sub>3</sub> genes, respectively; 43H, *T. elongatus* strain with a His-tag on the C terminus of CP43

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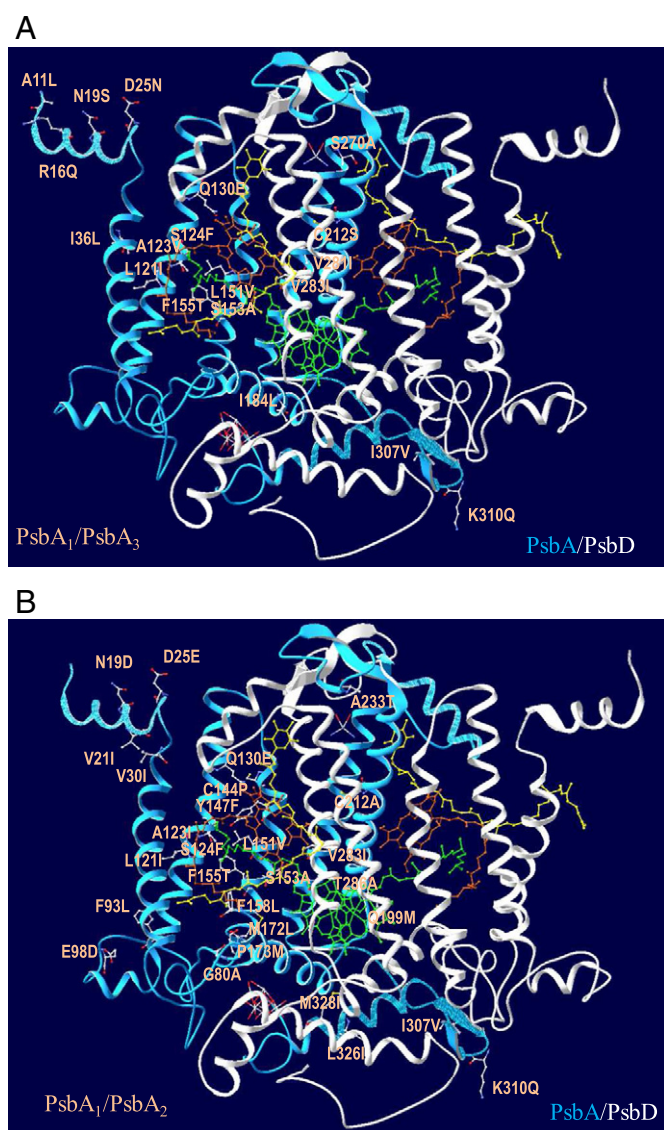
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cyanobacterial *psbA* gene family [14] and strategies for the *psbA* gene expression are available in the literature [21].

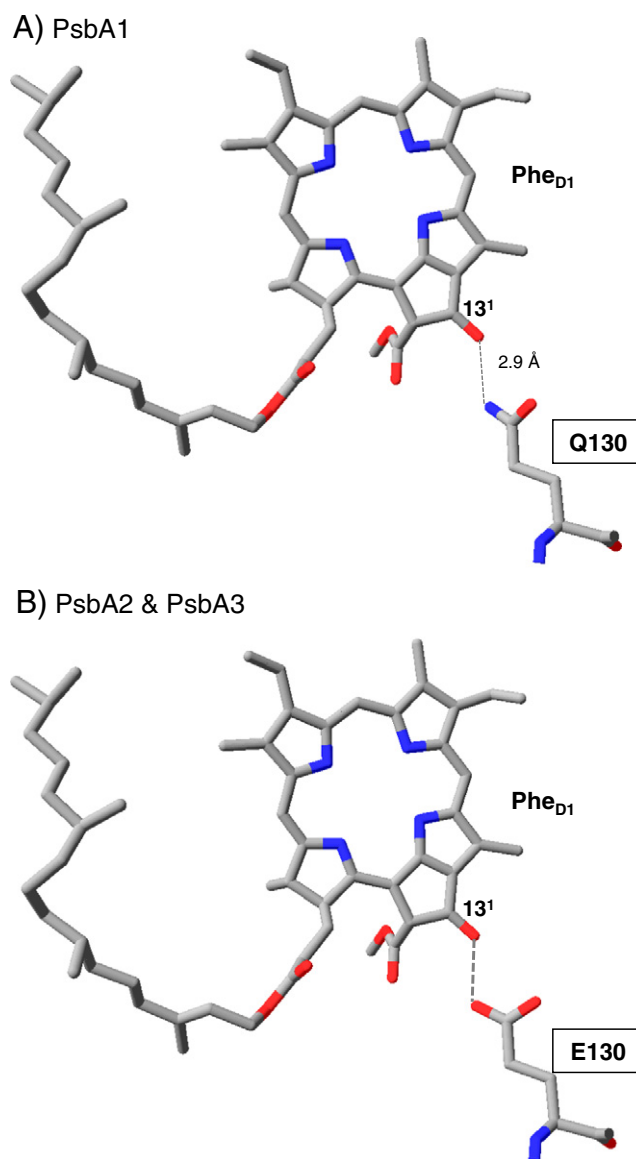
The thermophilic cyanobacterium *Thermosynechococcus elongatus* has three different *psbA* genes in its genome [22]. From the translated sequences and among the 344 residues of the PsbA proteins, 21 differ between PsbA1 and PsbA3, 31 between PsbA1 and PsbA2 and 27 between PsbA2 and PsbA3 (Fig. 1). The *psbA<sub>1</sub>* gene is constitutively expressed under “normal” laboratory conditions, while the transcription of *psbA<sub>3</sub>* occurred under high-light or UV light conditions [13,23,24]. The transcription of the *psbA<sub>2</sub>* gene has been reported to be, at least partially, induced under microaerobic conditions [15]. In contrast, with the case of the other cyanobacteria mentioned above, the differences in the D1 sequences in *T. elongatus* raise the possibility that the regulation at the transcription level is not a mere adjustment of the protein synthesis but rather an acclimation at the functional level whereby the functional properties of PSII are adjusted to cope with the increased photon flux. In this review, we summarize the changes in the properties of the redox cofactors depending on the D1 variant constituting PSII in *T. elongatus* that have been already identified.

## 2. The pheophytin, Phe<sub>D1</sub>

Among the amino acids which differ between PsbA1, PsbA2 and PsbA3, the residue at position 130 has caught much attention as shown in Figs. 1 and 2. Raman spectroscopy [25], EPR [26] and FTIR studies [27] have shown that it is H-bonded to the 13<sup>1</sup>-keto of Phe<sub>D1</sub>. In *T. elongatus* [27], the FTIR difference spectra of PsbA1-PSII exhibited the 13<sup>1</sup>-keto C=O bands at 1682 and 1605 cm<sup>-1</sup> in Phe<sub>D1</sub> and Phe<sub>D1</sub><sup>-</sup>, respectively, while the corresponding bands in PsbA3-PSII were observed at frequencies lower by 1–3 and 18–19 cm<sup>-1</sup>, respectively. This larger frequency shift in Phe<sub>D1</sub><sup>-</sup> than Phe<sub>D1</sub> by the change of the H-bond donor was well reproduced by density functional theory calculations for the Phe models H-bonded. Thus, the substitution of a glutamine in PsbA1-PSII for a glutamate in PsbA2-PSII and PsbA3-PSII results in a stronger H-bond with, as a consequence, an expected less negative  $E_m(\text{Phe}_{D1}/\text{Phe}_{D1}^-)$  in PsbA3-PSII than in PsbA1-PSII [28,29]. Such a change is therefore expected to modulate the energy level of the  $\text{P}_{680}^+\text{Phe}_{D1}^-$  radical pair, the free energy change associated with charge separation being larger with Glutamate than with Glutamine.



**Fig. 1.** Amino acid differences between either PsbA1 and PsbA3 (A) or PsbA1 and PsbA2 (B). Blue helices and white helices belong to D1 and D2, respectively. Numbers correspond to the amino acid sequence of D1. Letters are amino acids of PsbA1, PsbA3 and PsbA2.



**Fig. 2.** Structures around Phe<sub>D1</sub> in PsbA1 (A) and PsbA2 and PsbA3 (B). The 13<sup>1</sup>-keto C=O of Phe<sub>D1</sub> gives weakly H-bond to Gln130 of PsbA1 with the distance of 2.9 Å. This residue is substituted by Glu in both PsbA2 and PsbA3. In those PSII, the hydrogen bond between the 13<sup>1</sup>-keto C=O of Phe<sub>D1</sub> and Glu130 was stronger than that in PsbA1-PSII [27].

A clear and consistent view has emerged on the energetic consequences of the D1-Q130E substitution in *Synechocystis* PCC 6803 [30–32] or the D1-E130Q substitution in *Chlamydomonas reinhardtii* [33,34]. However, the increase of the redox potential of Phe<sub>D1</sub> in *T. elongatus* was found to be only 17 mV from –522 mV in PsbA1-PSII [28] to –505 mV in PsbA3-PSII [29]. This increase was half the one observed upon single site directed mutagenesis in *Synechocystis* PCC 6803 [30–32]. This led us to propose that the effects of the D1-Q130E substitution could be, at least partly, compensated for by some of the additional amino-acid changes associated with the PsbA3 for PsbA1 substitution [16,29]. Thermoluminescence and fluorescence studies of PsbA1- and PsbA3-containing PSII from *T. elongatus* have shown that, although qualitatively consistent with the effect of the D1/Q130E point mutation, the consequences of the PsbA1 for PsbA3 substitution on the kinetics and thermodynamic characteristic of the  $S_2Q_A^{\cdot-}$  charge recombination were also significantly less pronounced than in *Synechocystis* PCC 6803 [13,29,35]. This suggested that the physiologically relevant shift from PsbA1 to PsbA3 does not sum-up to the D1-Q130E change and that some (or all) of the 20 additional amino-acid substitutions contribute to determine the overall functional properties of the PsbA3 containing PSII. This suggestion found recently a strong support. Indeed, by studying the  $S_2Q_A^{\cdot-}$  charge recombination in the presence of DCMU by thermoluminescence and fluorescence in a PsbA3/E130Q single site directed mutant [36] we found that the  $E_m(\text{Phe}_{D1}/\text{Phe}_{D1}^{\cdot-})$  decreased by  $\approx 30$ –35 mV, a shift with a similar amplitude, but of course in an opposite direction, to that observed in *Synechocystis* PCC 6803 [13,30–32] and similar both in terms of direction and amplitude to that observed in *C. reinhardtii* [33,34]. The E130Q mutation in PsbA3 also hardly affected the quantum efficiency as characterized by the yield of  $S_2Q_A^{\cdot-}$  formation and the period four oscillations [36]. This is in agreement with previous reports [30,37] in which the Q130E change in *Synechocystis* PCC 6803 barely affected the nanosecond quantum yield of radical pair formation.

The  $E_m(\text{Phe}_{D1}/\text{Phe}_{D1}^{\cdot-})$  has not yet been determined in PsbA2-PSII. In PsbA2-PSII, this amino acid of position 130 is also a glutamine like PsbA3-PSII. However, due to the compensatory effects evoked above, only a direct measurement by spectro-electrochemistry like that one done in PsbA1-PSII and PsbA3-PSII is expected to give us an accurate value in this variant. However, indirect observations like the electrochromic blue shift of the Phe<sub>D1</sub> Q<sub>x</sub> absorption around 545 nm undergone by Phe<sub>D1</sub> upon reduction of Q<sub>A</sub> (and known as the C-550 bandshift) are already available. The C-550 bandshift was found to be red shifted by  $\approx 3.0$  nm for the PsbA3-PSII sample, relative to the PsbA1-PSII sample, as expected. This is due to the stronger H-bond to the <sup>13</sup>C-keto of the Phe<sub>D1</sub> from the carboxylate group of Q130 in PsbA3-PSII than from Q130 in PsbA1-PSII. In the PsbA2-PSII sample the electrochromic bandshift was found to be similar to that in the PsbA3-PSII sample which suggests that the electrostatic interaction which triggers this electrochromic bandshift is similar in PsbA2-PSII and PsbA3-PSII [36].

### 3. The primary quinone, Q<sub>A</sub>

Measuring the fluorescence intensity against the electrode potential resulted in a value for  $E_m(Q_A/Q_A^{\cdot-})$  of  $\approx -140$  mV in PsbA1-PSII [38] and  $\approx -102$  mV in PsbA3-PSII [29]. At first sight, this shift in  $E_m(Q_A/Q_A^{\cdot-})$  may seem unexpected because Q<sub>A</sub> is bound to PsbD (D2) rather than PsbA (D1). However, Q<sub>A</sub> is linked to the Q<sub>B</sub> binding site, made by the D1 protein, through the Q<sub>A</sub>–H214(D2)–Fe–H215(D1)–Q<sub>B</sub> molecular bridge. A FTIR study together with docking calculations [39] (see also [40]) suggested that the H-bond strength between D1/H215 and Q<sub>B</sub> influences the H-bond strength between D2–H214 and Q<sub>A</sub> through this molecular bridge, so that any change in the Q<sub>B</sub> site may propagate through this H-bond wire to Q<sub>A</sub> and possibly may lead to a shift of  $E_m(Q_A/Q_A^{\cdot-})$ .

Such a study has not yet been performed in PsbA2-PSII. However, by following the charge recombination in the light-induced radical pair  $S_1\text{Tyr}_Z^{\cdot-}Q_A^{\cdot-}$  at cryogenic temperatures by time-resolved EPR for different configurations of PSII that are expected to affect the driving force of the reaction (oxidation states  $S_0$ ,  $S_1$  or  $S_2$  of the Mn<sub>4</sub>CaO<sub>5</sub> cluster; PsbA1, PsbA2 or PsbA3 as D1 protein) similar  $S_1\text{Tyr}_Z^{\cdot-}Q_A^{\cdot-}$  charge recombinations were observed in PsbA2-PSII and PsbA3-PSII [41]. Since, as it will be discussed below, the  $E_m(\text{Tyr}_Z^{\cdot-}/\text{Tyr}_Z)$  value seems not affected by the PsbA exchange in the  $S_1$ -state, this predicts that the  $E_m(Q_A/Q_A^{\cdot-})$  value in PsbA2-PSII is likely similar to that in PsbA3-PSII.

### 4. The secondary quinone, Q<sub>B</sub>, and the non-heme iron

Little is known on the properties of the secondary quinone Q<sub>B</sub> that could vary depending on the D1 variant. Switching from PsbA1 to PsbA3 results in an amino acid substitution at position 270 (a serine in PsbA1 and an alanine in PsbA3 and PsbA2) as shown in Fig. 3 and it has been proposed that this might modify the structure of the Q<sub>B</sub> site by changing the H-bond strength with D1–H215, and hence this might shift the  $E_m(Q_A/Q_A^{\cdot-})$  value [42]. As mentioned in [43], the D1–S270A substitution may also contribute to loosen the H-bond with the head group of a lipid sulfoquinovosyldiacylglycerol (SQDG) located in the Q<sub>B</sub> site and may be the rationale behind the difference of the binding characteristics of herbicides such as DCMU or bromoxynil to the Q<sub>B</sub> site [16,43].

It has been observed that the addition of DCMU to PSII in the  $Q_A\text{Fe}^{\text{II}}Q_B^{\cdot-}$  state induces the formation of the  $Q_A\text{Fe}^{\text{III}}Q_B$  state in a fraction of centers, this fraction being larger in PsbA3-PSII than in PsbA1-PSII [43]. This observation could be explained in the framework of a square thermodynamic diagram in which the  $E_m$  of the Fe<sup>II</sup>/Fe<sup>III</sup> couple in the presence and in the absence of the herbicide differs if indeed the binding constant of DCMU depends on the redox state of the non-heme iron. In this framework, the  $E_m$  of the non-heme iron was proposed to be slightly more positive in PsbA1-PSII than in PsbA3-PSII [43].

The limiting step of the overall oxygen evolution *in vitro* is the exchange of the doubly reduced Q<sub>B</sub> molecule by an oxidized one. From an activity of  $\approx 6000 \mu\text{mol O}_2 (\text{mg Chl})^{-1} \text{h}^{-1}$ , in PsbA3-PSII (*i.e.*  $\approx 50 \text{ O}_2$  molecules per second), one can infer for the limiting step a  $t_{1/2}$  of 20 ms, which corresponds to 2 Q<sub>B</sub>/Q<sub>B</sub>H<sub>2</sub> exchanges. In both PsbA1-PSII and PsbA2-PSII the O<sub>2</sub> activity is generally found close to  $\approx 3000$ –4000  $\mu\text{mol O}_2 (\text{mg Chl})^{-1} \text{h}^{-1}$  [29,44]. This could suggest a slower Q<sub>B</sub>/Q<sub>B</sub>H<sub>2</sub> exchange in PsbA1-PSII and PsbA2-PSII than in PsbA3-PSII. In PsbA2-PSII, the oxygen evolving rate is about 75% of that of PsbA1-PSII. Since the amino-acid sequence around Q<sub>B</sub> of PsbA2-PSII is similar to that of PsbA3-PSII further experiments will be required to understand the reasons of this apparent contradiction.

### 5. Electron transfer from Tyr<sub>Z</sub> to P<sub>680</sub><sup>+</sup>

The amino acid sequences around the Mn<sub>4</sub>CaO<sub>5</sub> cluster are similar for PsbA1-PSII, PsbA2-PSII and PsbA3-PSII. Expectedly, the rate of the  $S_3$  to  $S_0$  transition ( $t_{1/2} \approx 1$ –2 ms) is similar in the 3 variants [44–46]. In contrast, in Mn-depleted PSII and at pH 9.2, the electron transfer rate from Tyr<sub>Z</sub> to P<sub>680</sub><sup>+</sup> was found two fold faster in PsbA3 than in PsbA1 [16]. The faster reduction rate of P<sub>680</sub><sup>+</sup> in PsbA3-PSII likely stems from a faster intrinsic electron transfer rate constant between the tyrosinate and P<sub>680</sub><sup>+</sup> and therefore may reflect a difference in the free energy change associated with the electron transfer from Tyr<sub>Z</sub> to P<sub>680</sub><sup>+</sup> and thus a different redox potential of either of these two players. We previously proposed that a hint in favor of a change in the redox potential of the P<sub>680</sub><sup>+</sup>/P<sub>680</sub> couple was the observation that the slow component in the reduction of P<sub>680</sub><sup>+</sup>, which develops in hundreds of microsecond time range and is usually assigned to the charge recombination between P<sub>680</sub><sup>+</sup> and Q<sub>A</sub><sup>•−</sup> [47,48] was also faster in PsbA3-PSII than in PsbA1-PSII. However, considering the above results that both the  $E_m(Q_A/Q_A^{\cdot-})$  and  $E_m(\text{Phe}/\text{Phe}^{\cdot-})$  values also differ between the two types of PSII, this observation cannot be unequivocally interpreted



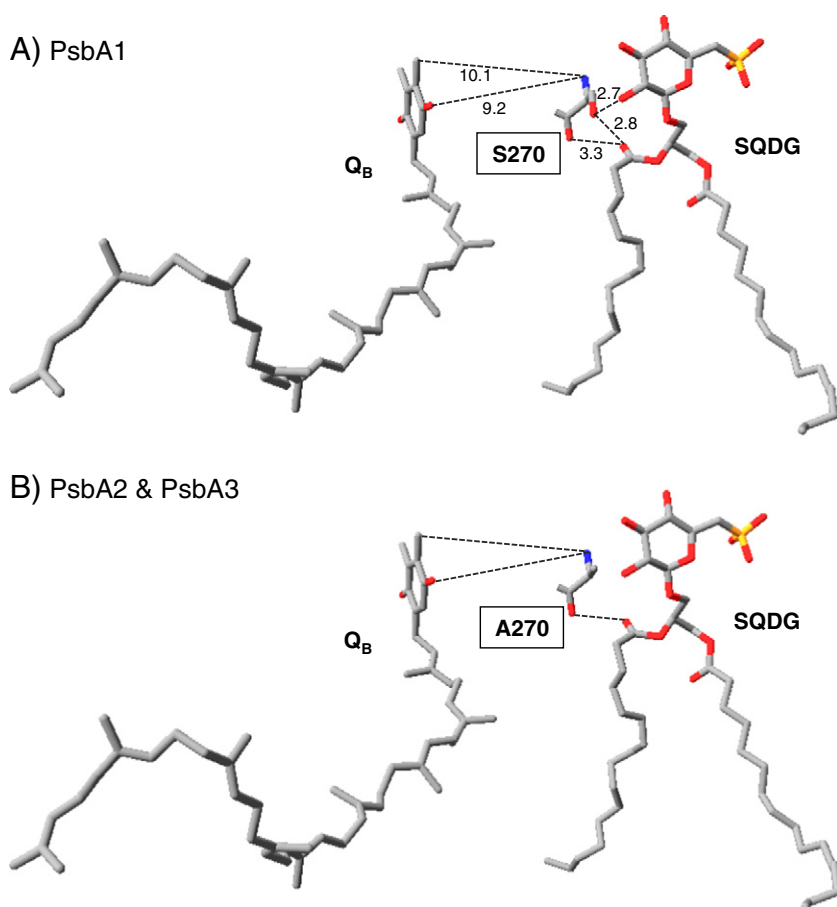


Fig. 3. Structure around SQDG, Ser270 and  $Q_B$  in PsbA1 (A). In PsbA2 and PsbA3, the Ser270 is substituted by Ala (B).

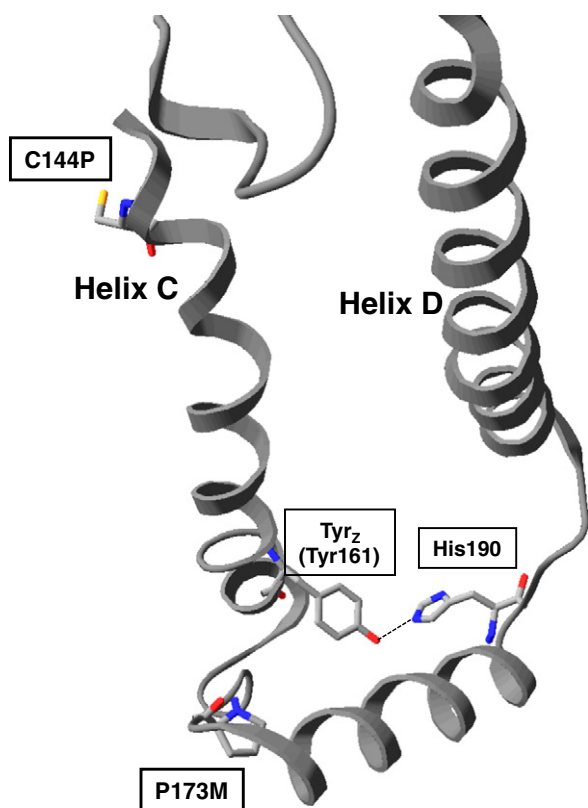
as reflecting a modification of the redox properties of  $P_{680}$  in Mn-depleted PSII.

In  $O_2$  evolving PSII, the electron transfer rate from  $Tyr_Z$  to  $P_{680}^{+}$  was found similar in PsbA3-PSII, e.g. [44], and in PsbA1-PSII, e.g. [49], whatever the S-state transition. In PsbA2-PSII, in the  $S_1$ -state, both the tens of ns and the tens of  $\mu$ s phases were found comparable to those in PsbA1/A3-PSII in terms of amplitude and  $t_{1/2}$  [44]. However, in the  $S_2$ - and  $S_3$ -states the reduction of  $P_{680}^{+}$  by  $Tyr_Z$  was much slower in PsbA2-PSII, particularly in hundreds of  $\mu$ s time domain. The finding that the reduction kinetics of  $P_{680}^{+}$  is hardly affected in the  $S_1$ -state shows that possible changes of the properties of the electron acceptor side originating from the PsbA(1/3) to PsbA2 substitution did not significantly increase the percentage of centers in which the  $P_{680}^{+}Q_A^{-}$  charge recombination occurred, at least with  $Q_B$  in the oxidized state (the experiment was performed in the presence of PPBQ). Therefore, it seems very unlikely that the slower  $P_{680}^{+}$  reduction kinetics in hundreds of  $\mu$ s time domain and detected in the  $S_2$  and  $S_3$  states originate from a charge recombination that would be more efficient in PsbA2-PSII. According to the current understanding of the multiphasicity of the reduction of  $P_{680}^{+}$  the ns components are kinetically limited by the electron transfer process, whereas the  $\mu$ s phases involve proton-coupled transfer reactions, e.g. [50–52]. In this framework, the results mentioned above would thus point to a slower proton transfer process in PsbA2-PSII. As shown in Fig. 4, two amino acid substitutions on the electron donor side of PsbA2-PSII may affect the orientation of the helices which respectively bear H190 and  $Tyr_Z$ : the C144P and P173M exchanges. These two substitutions may impact the H-bond between  $Tyr_Z$  and H190 and/or the H-bond network in which these two residues are involved. If such is indeed the case, this would be expected to affect the rates of the proton transfer steps associated with the oxidation of  $Tyr_Z$ . EPR spectroscopy, which has been shown to probe the geometry and the environment

of the  $Tyr_Z$  phenol ring, e.g. [53,54], indeed revealed a change, in PsbA2-PSII, in the split EPR signal attributed to the  $(S_2Tyr_Z)^{\cdot}$  state formed by the NIR-induced conversion of the manganese cluster, in the  $S_3$ -state, into an “activated” state able to oxidize  $Tyr_Z$  and thus leading to the formation of  $(S_2Tyr_Z)^{\cdot}$  at the expense of the  $S_3Tyr_Z$  state [55].

## 6. Electron transfer from the $Mn_4CaO_5$ cluster to $Tyr_Z$

The electron transfer rates between the  $Mn_4CaO_5$  cluster and  $Tyr_Z$  have been essentially probed by following the absorption changes at 292 nm. Absorption changes at this wavelength reflect the  $Mn_4CaO_5$  cluster valence changes and the  $Tyr_Z$  redox state changes occurring in the  $S_1Tyr_Z$  to  $S_2Tyr_Z$ ,  $S_2Tyr_Z$  to  $S_3Tyr_Z$ ,  $S_3Tyr_Z$  to  $S_0Tyr_Z$ , and  $S_0Tyr_Z$  to  $S_1Tyr_Z$  transitions [56]. No significant differences were found between the 3 PSII variants [44–46]. It was pointed out that the observation that the PsbA1/3 to PsbA2 exchange affects the  $\mu$ s components in the oxidation of  $Tyr_Z$  by  $P_{680}^{+}$  while keeping unaffected the proton release associated with the  $S_3Tyr_Z$  to  $S_0Tyr_Z$  transition suggests that this particular proton release does not originate from the same H-bond network as the one involved in the proton transfer triggered by the formation of  $Tyr_Z^{\cdot}$  [44]. The latter has been described as a sequence of push–pull steps that would be initiated by the transfer of the phenolic proton from  $Tyr_Z$  to N $\epsilon$  of H190. The identity of the “proton releaser” during the  $S_3Tyr_Z$  to  $S_0Tyr_Z$  is not known, and several candidates have been considered. A substrate water molecule is an obvious one. Alternatively, it could be a protonated base, proposed to be CP43-R357 [57], that would undergo a  $pK_a$  shift upon the formation of the  $S_3Tyr_Z$ ... HN $\epsilon$ (H190) $^{+}$  state and would, by acting as a proton acceptor from water, promote water splitting. These different proton transfer events thus have essentially different mechanistic implications. Although one mainly reflects electrostatic relaxation, the other sets the stage for all



**Fig. 4.** Substitutions of Cys144Pro and Pro173Met on helix C in PsbA2. In both PsbA1 and PsbA3, these residues are Cys144 and Pro173. Hydrogen-bonded Tyr<sub>2</sub> and His190 are from the helix C and helix D. In PsbA2-PSII, distance of this hydrogen bond was longer than those in PsbA1- and PsbA3-PSII [44].

the players in the water-splitting process. In such a framework, it is not surprising that they involve different molecular actors, and these results support this expectation. Notably, they also point to a necessary conformational change to account for the fact that a new proton releaser that had stayed inactive until the formation of  $S_3$  would come into play when  $S_3\text{Tyr}_2\cdots\text{HN}\epsilon(\text{H190})^+$  is formed.

## 7. Differences in the interactions between the subunits of Photosystem II dependant on D1 protein variants

The structural stability of PSII upon a PsbA1/PsbA3 exchange has been investigated by studying the effects of PsbJ deletion in strains expressing either PsbA1 or PsbA3 [42]. The PsbJ subunit is a 4-kDa transmembrane polypeptide that is surrounded by PsbA, PsbK, and cytochrome  $b_{559}$  (Cyt  $b_{559}$  = PsbE + PsbF + heme). The PsbA3/ $\Delta$ PsbJ-PSII was not significantly affected. Indeed, the polypeptide contents as revealed by SDS polyacrylamide gel electrophoresis and MALDI-TOF mass spectroscopy, the Cyt  $b_{559}$  properties, and the proportion of PSII dimer were similar to those found for PsbA3-PSII. In contrast, in 43H/ $\Delta$ PsbJ-PSII (a strain having the His6-tag on CP43 and expressing PsbA1) the stability of the dimer was greatly diminished, the EPR properties of the Cyt  $b_{559}$  likely indicated a decrease in its redox potential, and many other PSII subunits were lacking. These results showed that most of the amino acid substitutions between PsbA1 and PsbA3, which could appear to be mainly conservative, must include side chains that are involved in a network of interactions between PsbA and the other PSII subunits. The 3D structure of PsbA3-PSII and PsbA2-PSII with a resolution comparable to that obtained for PsbA1-PSII will help us to identify these interactions.

Interestingly, it was recently reported that the deletion of PsbJ in a strain similar to our 43H/ $\Delta$ PsbJ-PSII, i.e. a strain which possesses the 3

*psbA* genes, induced the expression of PsbA3 instead of PsbA1 [58]. In our conditions, we have found that the C-550 band-shift in 43H/ $\Delta$ PsbJ-PSII was similar to that one in PsbA1-PSII (Sugiura, Rappaport, Boussac, unpublished). This result shows clearly that under our conditions the PsbA1/PsbA3 exchange did not occur upon the deletion of PsbJ. It seems likely that different culture conditions could be at the origin of such differences. They point out the importance of the culture conditions in the interpretation of the data.

## 8. PsbA1-PSII and PsbA3-PSII under photoinhibitory conditions

The sensitivity to high light illuminations of PSII with either PsbA1 (WT\*1 cell strain) or PsbA3 (WT\*3 cell strain) as the D1 protein have been found to be different [24,59]. It should be noted that in our work [59] the WT\*1 strain had only the *psbA<sub>1</sub>* gene whereas in [24] the strain expressing PsbA1 contained both the *psbA<sub>1</sub>* and *psbA<sub>2</sub>* genes. When the cells were cultivated under high light conditions the following results were found: (i) the  $\text{O}_2$  evolution activity decreased faster in WT\*1 cells than in WT\*3 cells both in the absence and in the presence of lincomycin, a protein synthesis inhibitor; (ii) in WT\*1 cells, the rate constant for the decrease of the  $\text{O}_2$  evolution activity was comparable in the presence and in the absence of lincomycin; (iii) the D1 content revealed by Western blot analysis decayed similarly in both WT\*1 and WT\*3 cells and much slowly than  $\text{O}_2$  evolution; (iv) the faster decrease in  $\text{O}_2$  evolution in WT\*1 than in WT\*3 cells correlated with a much faster inhibition of the  $S_2$ -state formation; and (v) the shape of the WT\*1 cells was altered. All these results were in agreement with a photo-inhibition process resulting in the loss of the  $\text{O}_2$  activity much faster than the D1 turnover in PsbA1-PSII and likely to a greater production of reactive oxygen species under high light conditions in WT\*1 than in WT\*3.

In *Synechocystis* 6803 it has been found that the D1/Q130E mutant showed a decreased  $^1\text{O}_2$  production concomitantly with a decreased rate of photodamage relative to the WT, whereas both  $^1\text{O}_2$  production and photodamage were enhanced in the D1/Q130L mutant [60]. From these results and those discussed above, we expected that in *T. elongatus* the WT\*3-E130Q cells would exhibit greater resistance to high light conditions than the WT\*3 cells. Although the growth rates of WT\*3-E130Q cells were slightly faster than WT\*3 cells even with a light intensity equal to  $80 \mu\text{mol}$  of photons  $\text{m}^{-2} \text{s}^{-1}$ , the resistance under a high light intensity ( $800 \mu\text{mol}$  of photons  $\text{m}^{-2} \text{s}^{-1}$ ) was hardly affected [36].

The data in [36] and [60] can yet be reconciled. Firstly, the light intensity used may differ in the various experiments reported in the literature. This will influence the steady state concentration of  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  vs.  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  and thus the sensitivity of the  $^1\text{O}_2$  yield on the E130Q mutation. At higher light intensities  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  is accumulated, the energy gap between  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  and  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  has little consequence on the  $^1\text{O}_2$  yield, whereas the  $E_\text{m}$  of  $\text{Phe}/\text{Phe}^{\cdot-}$  will influence the direct recombination rate in competition with the triplet route. Under lower light intensities, both the energy gap between  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  and  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  and the  $E_\text{m}$  value of  $\text{Phe}/\text{Phe}^{\cdot-}$  matter. If we consider that the accumulated state is  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$ , the benefit of a larger energy gap between  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  and  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  (41 mV) in PsbA3/E130Q-PSII compared to that in PsbA3-PSII (26 mV) is likely offset by the disadvantage of a much lower value by 30 mV for the  $E_\text{m}$  of  $\text{Phe}/\text{Phe}^{\cdot-}$ . The case where  $\text{Phe}_{\text{D1}}\text{Q}_\text{A}^-$  is photo-accumulated is unlikely because we would expect more photodamage something that is not observed. Finally, the fact that the downshift by 30 mV of the  $E_\text{m}$  of  $\text{Phe}/\text{Phe}^{\cdot-}$  in PsbA3/E130Q-PSII did not decrease the rate of the photodamage of the cells when compared to WT\*3 cells could suggest that in *T. elongatus* and in PsbA3-PSII the  $E_\text{m}$  of  $\text{Phe}/\text{Phe}^{\cdot-}$  is not the only important parameter in the protection against photodamage. It is possible that the other 20 amino acid changes occurring within the PsbA1 to PsbA3 swap could play a totally different protective role.

The response of WT\*1 and WT\*3 cells to environmental changes should take into account that the *psbA<sub>1</sub>* and *psbA<sub>3</sub>* genes have not the

same promoter. In [36], both the *psbA<sub>3</sub>* gene and the gene encoding PsbA3/E130Q gene have the same promoter so that the effects of the E130 to Q130 substitution on the sensitivity to high light conditions were not complicated by different PsbA expressions depending on the environment conditions. Similarly, by using a heterologous expression system of the two D1 isoforms of PsbA of *Synechococcus elongatus* PCC 7942 (with D1:1 expressed under low light conditions and D1:2 up-regulated in high light or stress conditions) in the green alga *C. reinhardtii*, functional advantages of D1:1-PSII and D1:2-PSII at low and high light regimes, respectively, are revealed [61].

In our work [36,59] and in [24], the faster recovery under low light intensity of the O<sub>2</sub> activity in WT\*3 cells than in WT\*1 cells after an exposure of the cells to strong light intensity is difficult to rationalize with the observation that under the same low light intensity the PSII in WT (or 43-H) cells is constituted of PsbA1. These data could be rationalized, for example, by a model in which (i) the expression of the *psbA<sub>3</sub>* gene in 43H cells is down regulated by the expression of the *psbA<sub>1</sub>* gene and (ii) strong light intensities inhibit the transcription of *psbA<sub>1</sub>* which in turn cancels the down-regulation of the *psbA<sub>3</sub>* transcription.

We have observed that the lag phase of *psbA<sub>3</sub>*-deleted mutant cells which still therefore contained both the *psbA<sub>1</sub>* and *psbA<sub>2</sub>* genes was shorter than that of WT\*1 cells (that contain only the *psbA<sub>1</sub>* gene) under the illumination of 60 μmol photons m<sup>-2</sup> s<sup>-1</sup>. This result suggests that the *psbA<sub>2</sub>* gene present in the *psbA<sub>3</sub>*-deleted mutant in [24] could be possibly transcribed during the lag phase.

## 9. PsbA3 and high light conditions

The consequence of the increase of the  $E_m(Q_A/Q_A^{\cdot-})$  by ≈ 40 mV and of the  $E_m(\text{Phe}/\text{Phe}^{\cdot-})$  by ≈ 17 mV is an increase by 23 mV of the energy gap between  $\text{Phe}_{D1}^{\cdot-}Q_A$  and  $\text{Phe}_{D1}Q_A^{\cdot-}$  in PsbA3-PSII when compared to PsbA1-PSII. This indicates that under a light intensity which results in an accumulation of  $Q_A^{\cdot-}$  the thermal repopulation of the  $\text{Phe}_{D1}^{\cdot-}Q_A$  state from the  $\text{Phe}_{D1}Q_A^{\cdot-}$  state is more difficult in PsbA3-PSII than in PsbA1-PSII which is thus expected to favor the direct charge recombination between  $P_{680}^+$  and  $Q_A^{\cdot-}$ . Therefore, since formation of  $^3[P_{680}^+\text{Phe}_{D1}^{\cdot-}]$  may lead to the formation of  $^3P_{680}$ , this implies that PsbA3-PSII would be less prone to photo-damage than PsbA1-PSII. In addition, it has been suggested that the preferential expression of PsbA3 under strong light [24] resulted in a more efficient direct charge recombination between the donor side and the acceptor side thus also preventing the production of harmful  $^1O_2$  from  $^3P_{680}$  itself formed by charge recombination in the thermally repopulated  $^3[P_{680}^+\text{Phe}_{D1}^{\cdot-}]$  state [13], but see [62] for a discussion. This reasoning relied on the assumption that the charge recombination occurred in the inverted region of the Marcus curve with a  $E_m(\text{Phe}/\text{Phe}^{\cdot-})$  value and a  $E_m(Q_A/Q_A^{\cdot-})$  value higher in PsbA3-PSII than in PsbA1-PSII. We have shown that, at least at cryogenic temperatures, such inverted region in Photosystem II exists [41]. Finally, if the  $Q_B/Q_BH_2$

exchange is indeed faster in PsbA3-PSII than in PsbA1-PSII this gives a third advantage to PsbA3-PSII over PsbA1-PSII under high light conditions in which  $Q_A^{\cdot-}$  is accumulated. This list is likely not exhaustive. For example, it was also found that in the PsbA1-PSII the quantum efficiency of photo-induced oxidation of side-pathway donors was lower which also indicates an advantage for PsbA3 under high light conditions [63].

All these different expectations were numerically assessed by a modeling of the electron transfer reactions in PSII similar to that done previously by Moser et al. [64] who relied on the knowledge provided the X-ray structure of the distances between the different redox cofactors in PSII and applied their empirical ruler that allows electron transfer rates to be calculated when the distance between the electron donor and acceptor and the driving force of the reaction are known [65]. We have followed a similar approach and compared the time courses of the redox changes of the main cofactors in PsbA1-PSII, PsbA3-PSII and PsbA3/E130Q-PSII. It appeared that in the three cases the quantum efficiency of  $S_2Q_A^{\cdot-}$  formation was hardly affected which was consistent with the experimental results [36]. The simulated kinetics for  $S_2Q_A^{\cdot-}$  charge recombination agreed with the experimental results, i.e. a similar kinetics in PsbA1-PSII and PsbA3-PSII [16] and a much slower charge recombination in PsbA3/E130Q-PSII [36]. In the same theoretical work, the simulated formations of singlet oxygen that were found to be ~60%, ~20% and ~7% in PsbA1-PSII, PsbA3-PSII, and PsbA3/E130Q-PSII, respectively, qualitatively correlated with the differential sensitivity to high light for PsbA1 vs. PsbA3, e.g. [59], and with the lack of detectable difference between PsbA3 and PsbA3/E130Q [36].

## 10. PsbA2-PSII and Tll0287 expression

A new hemoprotein has been found to be expressed when the *T. elongatus* genome has only the *psbA<sub>2</sub>* gene for D1 [66]. This hemoprotein was found in both the fraction containing the purified PsbA2-PSII core complex and the non-membrane protein pool. From MALDI-TOF/TOF spectrometry, N-terminal sequencing and MALDI-MS/MS analysis upon tryptic digestion, the new hemoprotein was identified to be the *tll0287* gene product with a molecular mass close to 19 kDa. Until now, *tll0287* was registered as a gene encoding a hypothetical protein with an unknown function. From the amino acid sequence and the EPR spectrum the 5th and 6th axial ligands of the heme iron are the H145 and likely either the Y93, Y159 or Y165, respectively. The amino acid sequence of Tll0287 and conserved residues are shown in Fig. 5. Homologous genes to *tll0287* are found in several cyanobacteria. Since the expression of the *psbA<sub>2</sub>* gene has been shown to occur under micro-aerobic conditions, under our cultivation conditions, i.e. in the presence of ≈ 20% O<sub>2</sub>, the production of Tll0287 in WT\*2 cells could be the answer to an oxidative stress. Future works would have to solve this question.

```

1      11      21      31      41
MVRIFLMALLMASLWIQGSPAPLASANPEELGKVVTAIEQLDOMRIGLAS

51      61      71      81      91
TLEGGTSEPTLDTFKAVCAPVGKQAKEIAAANGWQVRQVALKYRNPNHAP*

101     111     121     131     141
RTALDVQALNQFDNNHHLQAFWQTDKEGVHYFRRIDVQASCLACHGAKNR

151     161     171     181     191
RPAFIQEKYPSDRAYGFRVGDRLRGMYAVTIPQIQALQTS*P

```

**Fig. 5.** Deduced amino acid sequence from the *tll0287* gene in *T. elongatus*. The underlined amino acid residues are those belonging to the main N-terminal sequences detected by using the Edman procedure. The frame shows the conserved heme-binding motif (CxxCH). The stars show the conserved tyrosine residues which are potential candidates to be the 6th heme iron axial ligand [66]. The doubly underlined amino acids are those belonging to the fragments identified with Mascot after tryptic digestion and MALDI-MS/MS.



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